

Available online at www.sciencedirect.com

Tetrahedron Letters

Tetrahedron Letters 49 (2008) 870–872

a-Amino 1,3-dithioketal mediated asymmetric synthesis of piperidines (L-733,060) and tetrahydrofuran glycines

Franklin A. Davis *, Tokala Ramachandar

Department of Chemistry, Temple University, Philadelphia, PA 19122, USA

Received 2 November 2007; revised 16 November 2007; accepted 27 November 2007 Available online 4 December 2007

Abstract

Sulfinimine-derived α -amino 1,3-dithianes and α -amino carbonyl chiral building blocks, are utilized in asymmetric syntheses of (+)-(tetrahydrofuran-2-yl)glycine and the 2,3-disubstituted piperidine (+)-L-733,060. - 2007 Elsevier Ltd. All rights reserved.

Enantiomerically pure α -amino aldehydes and ketones, generally prepared from α -amino acids, are widely used chiral building blocks for asymmetric synthesis.^{[1](#page-2-0)} However, a-amino carbonyl compounds are notoriously unstable and rapidly epimerize and self-condense even when suitably Nprotected. Many of these problems can be avoided by using N -sulfinyl α -amino 1,3-dithianes 1, new sulfinimine-derived chiral building blocks for a-amino aldehyde, and ketone synthesis (Scheme 1).^{[2,3](#page-2-0)} These building blocks are readily

Corresponding author. Tel.: +1 215 204 0477; fax: +1 215 204 0478. E-mail address: fdavis@temple.edu (F. A. Davis).

0040-4039/\$ - see front matter © 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2007.11.170

prepared in high diastereomeric purity by addition of 2-lithio-1,3-dithianes to sulfinimines. 4 Acid hydrolysis affords the enantiomerically pure free amine, leaving the carbonyl group protected, resulting in unique opportunities for functional group manipulation. We employed these α amino 1,3-dithianes in highly diastereoselective asymmetric syntheses of functionalized prolines including $(-)$ -3hydroxy-3-methyl proline $(2)^2$ $(2)^2$ $(2)^2$ and $(-)$ -2,3-trans-3,4-cisdihydroxyproline (3) (3) (3) .³ As an extension of this methodology, we describe the concise asymmetric synthesis of the 2,3-disubstituted piperidine $(2S,3S)$ -(+)-L-733,060 (4) and the unnatural constrained α -amino acid (+)-(2R,2'S)-(+)-2-(tetrahydrofuran-2-yl)glycine (THFG 5) from a common intermediate.

The key strategy employed in our synthesis of hydroxy prolines $(-)$ -2 and $(-)$ -3 was the cyclization of a 2hydroxyethyl moiety (R') in 1 to form the pyrrolidine ring. Cyclization of a 3-hydroxypropyl group (R') in 1 would produce a piperidine ring. Addition of 2-lithium-1,3-dithiane 7^5 7^5 to $(S)-(+)$ -N-(benzylidene)-p-toluenesulfinamide ([6](#page-2-0))⁶ gave *N*-sulfinyl α -amino 1,3-dithiane (S_S,S)-(+)-8 in 94% de and 81% yield of the major diastereoisomer [\(Scheme 2](#page-1-0)). Hydrolysis of $(+)$ -8 with 1,3-dibromo-5,5dimethylhydantoin (9) in aqueous acetone not only hydrolyzes the thioketal group, but also oxidizes the N-sulfinyl group to the N-tosyl protecting group affording (S) - $(-)$ -10 in 73% yield for the one pot sequence. Reduction of the amino ketone with N a BH ₄ gives the expected syn

alcohol $(1S,2S)$ -(+)-11 $(dr = 10:1)$ in 87% yield of the major diastereoisomer. Selective tosylation of the primary alcohol in $(+)$ -12 was readily accomplished with TsCl/pyridine affording $(1S,2S)-(+)$ -13 in 87% yield.

Next, treatment of amino alcohol $(+)$ -13 with Et₃N gave the expected, kinetically favored, furan $(+)$ -14 in 83% yield (Scheme 3). Oxidation of the phenyl group in $(+)$ -14 with RuCl3/NaIO4 gave the carboxylic acid 15, which was

isolated as the methyl ester with TMSCHN₂ affording $(2R,2'S)-(+)$ -5 in 54% for the two steps.^{[7](#page-2-0)} Conformationally constrained unnatural amino acids such as (+)-5 are in demand for the synthesis of constrained peptide surrogates, which have increased potency, stability, bioavailability, and selectivity.

To prepare a piperidine from $(+)$ -11 required protection of the 2-hydroxy group as the OTBS ether. Treatment of (+)-16 with p-TSA at 0° C for less than 30 min made it possible to selectively deprotect the primary OTBS, affording $(+)$ -17 in 82% yield (Scheme 4). With TsCl/Et₃N, $(+)$ -17 gave piperidine $(+)$ -18 in 88% yield and the hydroxy piperidine $(+)$ -19 was obtained in 90% yield using TBAF. Etherification of the hydroxy group with NaH and 3,5 bis(trifluoromethyl)benzyl bromide gave (2S,3S)-(+)-20. Finally, *N*-tosyl deprotection with Na/liq. NH₃ at -78 °C afforded (2S,3S)-(+)-L-733,060 (4) in 78% yield (Scheme 4).^{[8,9](#page-2-0)} (+)-L-733,060 (4) is a potent neurokinin substance P receptor antagonist, which exhibits strong antiemetic activity.[10](#page-2-0)

In summary, the utility of α -amino 1,3-dithianes as chiral building blocks for the asymmetric synthesis of heterocycles has been demonstrated by the preparation of amino furan $(+)$ -THFG-5 and 2,3-disubstituted piperidine $(+)$ -L-733,060 (4) from common amino diol intermediate (+)- 11.^{[12](#page-2-0)} In particular, our synthesis of $(+)$ -4 in 11 steps under nine operations (18% overall yield) from sulfinimine $(+)$ -6 is one of the most concise to date. 2,3-Disubstituted piperidines are structural units found in natural products and

several drug candidates, 11 and the structural diversity of available sulfinimine-derived α -amino 1,3-dithianes make this protocol well suited for enantiomer and analog synthesis.¹⁰

Acknowledgment

This work was supported by grants from the National Institute of General Medicinal Sciences (GM57878 and GM51982).

References and notes

- 1. For reviews on the synthesis of chiral α -amino aldehydes and ketones see: (a) Jurczak, J.; Golebiowski, A. Chem. Rev. 1989, 89, 149–164; (b) Fisher, L. E.; Muchowski, J. M. Org. Prep. Proced. Int. 1990, 22, 399– 484; (c) Reetz, M. T. Chem. Rev. 1999, 99, 1121–1162; (d) Gryko, D.; Chalko, J.; Jurczak, J. Chirality 2003, 15, 514–541.
- 2. Davis, F. A.; Ramachandar, T.; Liu, H. Org. Lett. 2004, 6, 3393– 3395.
- 3. Davis, F. A.; Ramachandar, T.; Chai, J.; Skucas, E. Tetrahedron Lett. 2006, 47, 2743–2746.
- 4. For recent reviews on the chemistry of sulfinimines see: (a) Zhou, P.; Chen, B.-C.; Davis, F. A. Tetrahedron 2004, 60, 8003–8030; (b) Senanayake, C. H.; Krishnamurthy, D.; Lu, Z.-H.; Han, Z.; Gallou, I. Aldrichim. Acta 2005, 38, 93–104; (c) Morton, D.; Stockman, R. A. Tetrahedron 2006, 62, 8869–8905; (d) Davis, F. A. J. Org. Chem. 2006, 71, 8993–9003.
- 5. Wei, W.-G.; Yao, Z.-J. Tetrahedron 2003, 59, 6621–6625.
- 6. Davis, F. A.; Zhang, Y.; Andemichael, Y.; Fang, T.; Fanelli, D. L.; Zhang, H. J. Org. Chem. 1999, 64, 1403–1406.
- 7. For earlier asymmetric synthesis of THFG see: (a) Williams, R. M.; Sinclair, P. J.; Zhai, D.; Chen, D. J. Am. Chem. Soc. 1988, 110, 1547– 1557; (b) Horikawa, M.; Busch-Petersen, J.; Corey, E. J. Tetrahedron Lett. 1999, 40, 3843–3846; (c) Kazamaier, U.; Pahler, S.; Endermann, R.; Habich, D.; Kroll, H.-P.; Riedl, B. Bioorg. Med. Chem. 2002, 10, 3905–3913; (d) Jirgensons, A.; Marinozzi, M.; Pellicciari, R. Tetrahedron 2005, 61, 373–377.
- 8. For earlier asymmetric syntheses of $(+)$ -4 see: (a) Bhaskar, G.; Rao, B. V. Tetrahedron Lett. 2003, 44, 915–917; (b) Huang, P.-Q.; Liu, L.-X.; Wei, B.-G.; Ruan, Y.-P. Org. Lett. 2003, 5, 1927–1929; (c) Kandula, S. R. V.; Kumar, P. Tetrahedron: Asymmetry 2005, 16, 3579–3583; (d) Yoon, Y.-J.; Joo, J.-E.; Lee, K.-Y.; Kim, Y.-H.; Oh, C.-Y.; Ham, W.-H. Tetrahedron Lett. 2005, 46, 739–741; (e) Oshitari, T.; Mandai, T. Synlett 2006, 3395–3398; (f) Cherian, S. K.; Kumar, P. Tetrahedron: Asymmetry 2007, 18, 982–987.
- 9. For related 2,3-disubstituted piperidines see: (a) Liu, L.-X.; Ruan, Y.-P.; Guo, Z.-Q.; Huang, P.-Q. J. Org. Chem. 2004, 69, 6001–6009; (b) Lemire, A.; Grenon, M.; Pourashraf, M.; Charette, A. B. Org. Lett. 2004, 6, 3517–3520; (c) Takahashi, K.; Nakano, H.; Fujita, R. Tetrahedron Lett. 2005, 46, 8927–8930.
- 10. (a) Baker, R.; Harrison, T.; Swain, C. J.; William, B. J. EP 0 528,495A1, 1993; (b) Harrison, T.; Williams, B. J.; Swain, C. J.; Ball, R. G. Bioorg. Med. Chem. Lett. 1994, 4, 2545–2550.
- 11. For leading references see: Davis, F. A.; Zhang, Y.; Li, D. Tetrahedron Lett. 2007, 48, 7838–7840.
- 12. Selected data: (+)-8, mp 109-110 °C; $[\alpha]_D^{20}$ 30.8 (c 0.4, CHCl₃); (-)-10, mp 93–95 °C; $[\alpha]_D^{20}$ –161.4 (c 0.42, CHCl₃); (+)-11, $[\alpha]_D^{20}$ +26.5 (c 0.7, CHCl₃); (+)-12, $[\alpha]_{\text{D}}^{20}$ 44.7 (c 1.3, CHCl₃); (+)-13, $[\alpha]_{\text{D}}^{20}$ 16.8 (c 0.81, CHCl₃); (+)-14, $[\alpha]_D^{20}$ +11.5 (c 0.68, CHCl₃); (+)-5, $[\alpha]_D^{20}$
+34.4 (c 1.1, CHCl₃); (+)-16, $[\alpha]_D^{20}$ +27.1 (c 0.72, CHCl₃); (+)-17,
mp. 152.353.9C; $[\alpha]_2^{20}$ +25.1 (c 0.4, CHCl); (+) 18, $[\alpha]$ mp 152–153 °C; $[\alpha]_{D}^{20}$ +35.1 (c 0.4, CHCl₃); (+)-18, $[\alpha]_{D}^{20}$ +59.2 (c 0.32, CHCl₃); (+)-19, $[\alpha]_D^{20}$ +76.31 (c 0.38, CHCl₃); (+)-20, $[\alpha]_D^{20}$
+58.6 (c 0.53, CHCl₃); (+)-4, $[\alpha]_D^{25}$ +36.12 (c, 0.64, CHCl₃).